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Aromatic Ring Synthesis by 1,3-Michael-Claisen Annulation : Formation of Dihydrobenzofurans and Tetrahydrochromans from α -Methylene γ -Butyrolactone and δ -Valerolactone.

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Abstract: Substituted dihydrobenzofurans and tetrahydrochromans have been prepared from respectively α -methylene- γ -butyrolactone and δ -valerolactone *via* a 1,3-Michael-Claisen reaction from two 3-carbon units, 1,1-bis-(methylthio)-2-propanone or butanone, and an α -methylene lactone.

Formation of aromatic rings is of great importance in the synthesis of natural products and biologically active compounds¹. Construction of benzenoid aromatic systems from aliphatic sources via annulation reactions has the great advantage to control the position of substituents at the initial stage of the synthesis.

We describe in this paper the preparation of substituted dihydrobenzofurans and tetrahydrochromans from respectively α -methylene γ -butyrolactone and δ -valerolactone via a 1,3-Michael-Claisen annulation reaction as shown on the retrosynthetic scheme.

Our approach is based on the work of Kim $^{1b, 2}$ who showed that phenolic compounds could be readily made via an 1,3-Michael-Claisen annulation from two 3-carbon units, such as 1,1-bis-(methylthio)-2-propanone 1a (R=H), and an α , β -unsaturated ketone or an α -methylene lactone.

This methodology allowed us to describe, using new experimental conditions, short syntheses of highly substituted dihydrobenzofurans and tetrahydrochromans.

1,1-bis-(methylthio)-2-propanone 1a was readily prepared from t-butyl acetoacetate which was first disulfenylated with methylthiotosylate³ and decarboxylated in modified Krapcho conditions⁴ (without addition of salts like NaCN), (Scheme I).

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1.1-bis(methylthio)-2-butanone **1b** was obtained by condensation of the anion of bis(methylthio)-methane on ethyl propionate (Scheme II).

Scheme II

In refluxing a mixture of α -methylene- γ -butyrolactone⁵ **2**, the thioacetal **1a** and sodium hydride (2 eq) in CH₂Cl₂, we observed the formation of the Michael adduct as a mixture of the three possible enols. This mixture was directly cyclized in acetic acid to give the tetrahydrobenzofuran **3a** in 50% overall yield (Scheme III).

MeS SMe 1) NaH, CH₂Cl₂,
$$\Delta$$
 2) TsOH cat., PhH, Δ 4a, R = H, 90% 4b, R = Me, 46% 6b, R = Me, 36% 6b, R =

Aromatisation was carried out in presence of p-TsOH to give the dihydrobenzofurane 4a in 90% yield.

The formation of **4a** was readily improved in heating directly the Michael adduct in presence of p-TsOH affording a 2 steps synthesis of the dihydrobenzofurane **4a** in 90% yield.

Compound **4b** was obtained in similar conditions with 46% yield. The lower yield is due to the chromatographic purification.

Ozaki and Kim^{2a} hydrolyzed the thioacetal function of type 3 intermediates in order to get catechol derivatives with mercuric chloride or perchlorate in rather poor yields. We found that transacetalization with mercuric acetate in methanol to the corresponding acetal 5 proceeded with a much better yield.

Finally aromatization in boiling acetic acid afforded the corresponding catechol derivatives 6. Once again the stability of 6b led to a lower yield.

Finally it was shown that the addition of MeLi to 5b, followed by aromatisation in acetic acid yielded the methylated benzofuran 7b in 70% yield.

Substituted tetrahydrochromans were also prepared by the same procedure from α -methylene δ -valerolactone 6 8 (Scheme IV).

A few remarks concerning the syntheses of the tetrahydrochromans of Scheme IV must be pointed out. Tetrahydrochroman 10a, obtained in 80% yield (from NMR), was contaminated by about 20% of the starting thioacetal 1a which could not be separated by chromatography (same Rf). Therefore in this case it is necessary to make first the tetrahydrochroman derivative 9a (50%) in acetic acid and then get 10a by aromatisation with p-TsOH.

Scheme IV

On the other hand, the direct 2 steps synthesis of 10b was faster than usual, it was necessary to heat the reaction mixture with p-TsOH only 4h (16h in the other cases).

For the synthesis of 12b, the first 3 steps gave the expected products in high yields, but the last step, the aromatisation of the product 11b, gave only 36% yield due to the low stability of 12b on silicagel.

Finally, in the synthesis of 13b, the last step, the aromatisation step in acetic acid gave a small amount of the corresponding phenol (20%) which could not be separated from 13b by chromatography.

Experimental Section.

1,1-Bis(methylthio)-2-propanone, 1a

1) 2,2-bis(methylthio)-t-butyl-acetoacetate: DBU (2 eq, 25.13 mL, 168 mmol) and methyl thiotosylate 3 (2 eq, 34 g, 168 mmol) were successively added to a solution of t-butyl acetoacetate (1 eq, 13.3 g, 84 mmol) in DMF (300 mL). The mixture was heated at 80°C during 30 min, allowed to cool and poured onto cold water (800 mL). The aqueous phase was extracted with a 1/1 ether-hexane mixture (3 x 200 mL), the combined organic layers were washed with 200 mL of a 10% HCl solution and water (200 mL), dried over Na₂SO₄ and concentrated under vacuum to yield 2,2-bis(methylthio)-t-butyl-acetoacetate (19.9g, 79.8 mmol, 95%) as yellow cristals.

IR (CCl₄): 3020-2840 (C-H), 1720 (C=O). ¹H NMR (200 MHz, CDCl₃): δ = 1.51 (s, 9H, t-Bu), 1.98 (s, 6H, SCH₃), 2.37 (s, 3H, H-4). ¹³C NMR (50 MHz, CDCl₃): 11.82 (SCH₃), 25.29 (C-4), 27.44 (t-Bu), 75.85 (C-2), 83.94 (C-5), 165.20 (C-1), 195.07 (C-3)

Anal. Cald for C₁₀H₁₈O₃S₂: C, 47.97; H, 7.39. Found: C, 48.13; H, 7.25.

2) A solution of DMSO (100 mL), 2,2-bis(methylthio)-t-butyl-acetoacetate (1 eq, 19.9 g, 77 mmol) and water (1 eq, 1.43 mL, 77 mmol) was heated at 160° C during 4h. After cooling, the mixture was extracted with CH₂Cl₂ (3 x 100 mL) and the joined organic layers were washed with water (100 mL), dried over Na₂SO₄ and concentrated under vacuum. The oily residue was distilled in a bulb to bulb Büchi apparatus (bp = 60° C/0,1 mmHg, apparatus temperature) to yield a pale yellow oil (9.8 g, 65.4 mmol, 85%).

IR (CCl₄): 3020-2820 (C-H), 1720 (C=O). ¹H NMR (200 MHz, CDCl₃): δ = 2.08 (s, 6H, SCH₃), 2.36 (s, 3H, H-3), 4.37 (s,1H, H-1). ¹³C NMR (50 MHz, CDCl₃): 11.62 (SCH₃), 25.72 (C-3), 60..70 (C-1), 198.45 (C-2)

Anal. Cald for C5H10OS2: C, 39.97; H, 6.71. Found: C, 39.62; H, 6.71.

1,1-Bis(methylthio)-2-butanone, 1b

A solution of bis(methylthio)methane (2 eq, 5 mL, 48.9 mmol) in THF (100 mL) was cooled at -78°C and n-butyllithium (2 eq, 33.3 mL of a 1.47M solution in hexane, 48.9 mmol) was added dropwise. The resulting mixture was stirred at -78°C during 2 h and added via a canula to a solution of ethyl propionate (60 mL) in THF (100 mL) previously cooled to -78°C. Stirring at -78°C was maintained during 30 min and hydrolysis performed by adding sat. NH4Cl solution (100 mL), The mixture was extracted with CH2Cl2 (3 x 100 mL) and the joined organic layers were washed with water (100 mL), dried over Na₂SO₄ and concentrated under vacuum (12 mmHg to remove the solvents and then 48 hours under 0.5 mmHg to remove starting materials in excess). A pale yellow oil (3.98 g, 24.2 mmol, 98 %) was obtained.

IR (CCl₄): 3020-2820 (C-H), 1720 (C=O). ¹H NMR (200 MHz, CDCl₃): δ = 1.10 (t, 3H, J=7.3 Hz, H-4), 2.05 (s, 6H, SCH₃), 2.69 (q, 2H, J=7. 3 Hz, H-3), 4.37 (s, 1H, H-1). ¹³C NMR (50 MHz, CDCl₃): 7.87 (C-4), 12.23 (SCH₃), 32.03 (C-3), 60.30 (C-1), 201.92 (C-2).

Anal. Cald for C6H12OS2: C, 43.87: H, 7.36. Found: C, 43.86; H, 7.54.

General procedure for the preparation of α -methylthio- β -hydroxy compounds.

In a three-necked round bottomed flask equipped with a dropping funnel, a suspension of NaH (2 eq) in CH₂Cl₂ (1 mL/mmol NaH) was cooled at 0°C with an ice bath. Thioacetal **1a** or **1b** (1 eq) in CH₂Cl₂ (1 mL/mmol thioacetal) was added dropwise to the suspension. After stirring for 10 min at 0°C, α-methylenelactone **2** or **8** (1 eq) dissolved in CH₂Cl₂ (1 mL/mmol lactone) was added dropwise and the ice bath was replaced by a heating bath. The suspension was refluxed for 16 h, allowed to cool down and poured onto a 10% aqueous HCl solution (10 mL/mmol thioacetal). The organic layer was decanted, the aqueous layer extracted with CH₂Cl₂ and the joined organic layers were washed with brine, dried over Na₂SO₄ and concentrated under vacuum. The residue was dissolved in benzene (5ml/mmol thioacetal) and a catalytic amount of p-TsOH added. The resulting solution was refluxed with a Dean-Stark apparatus during xh. After cooling and distillation of benzene under vacuum, the product was submitted to flash-chromatography (see products description).

2,3-Dihydro-5-methylthio-6-hydroxy-benzofuran, 4a

Prepared from compounds 1a and 2. Refluxing time x = 120h. Chromatography: CH₂Cl₂. Yield: 90%, pale pink cristals, R_f: 0.50 (CH₂Cl₂), mp: 62-63°C.

IR (CCl4) : 3400 (OH), 3000-2860 (C-H). 1 H NMR (200 MHz, CDCl3) : δ = 2.25 (s, 3H, SCH3), 3.13 (t, 2H, J=8.5Hz, H-3), 4.58 (t, 2H, J=8.5Hz, H-2), 6.47 (s, 1H, H-7), 6.80 (s, 1H, OH), 7.29 (bs, 1H, H-4). 13 C NMR (50 MHz, CDCl3) : 20.60 (S-CH3), 28.67 (C-3), 72.04 (C-2), 96.40 (C-7), 110.70 (C-5), 119.46 (C-9), 130.85 (C-4), 158.93 (C-6), 162.55 (C-8).

Anal. Cald for C9H10O2S: C. 59.48; H. 5.57. Found: C. 59.32; H. 5.53.

2,3-Dihydro-6-methylthio-7-hydroxy-benzopyran, 10a

Prepared from compounds 8 and 1a. Refluxing time x = 72h. Yield: 80% (¹H NMR, could not be separated from 1a). Rf: 0.67 (CH₂Cl₂).

¹H NMR (200 MHz, CDCl₃): δ = 1.98 (tt, 2H, J₃₋₄=6.5Hz, J₃₋₂=5Hz, H-3), 2.26 (s, 3H, SCH₃), 2.71 (t, 2H, J=6.5Hz, H-4), 4.16 (t, 2H, J=5Hz, H-2), 6.44 (s, 1H, H-8), 6.58 (s, 1H, OH), 7.17 (bs, 1H, H-5)

2,3-Dihydro-5-methylthio-6-hydroxy-7-methyl-benzofuran, 4b

Prepared from compounds 1b and 2. Refluxing time x = 2.5h. Chromatography: CH₂Cl₂/hexane 20/80. Yield: 46%, pale yellow oil, $R_f: 0.51$ (CH₂Cl₂/hexane 50/50).

1. IR (CCl4) : 3400 (OH), 3000-2860 (C-H). 1 H NMR (200 MHz, CDCl3) : δ = 2.14 (s, 3H, CH3), 2.25 (s, 3H, S-CH3), 3.15 (t, 2H, J=8.5Hz, H-3), 4.58 (t, 2H, J=8.5Hz, H-2), 6.89(s, 1H, OH), 7.17 (bs, 1H, H-4). 13 C NMR (50 MHz, CDCl3) : δ = 20.92 (S-CH3), 29.49 (C-3), 71.82 (C-2), 106.43 (C-7), 110.60 (C-5), 118.36 (C-9), 127.84 (C-4), 155.15 (C-6), 160.95 (C-8)

Anal. Cald for C₁₀H₁₂O₂S: C. 61.20; H, 6.16. Found: C, 61.11; H, 6.33.

2,3-Dihydro-6-methylthio-7-hydroxy-8-methyl-benzopyran, 10b

Prepared from compounds 8 and 1b. Refluxing time x = 4h. Chromatography: CH₂Cl₂. Yield: 68%, a pale yellow oil, Rf: 0.95 (CH₂Cl₂).

IR (CCI₄): 3400 (OH), 3000-2840 (C-H). 1 H NMR (200 MHz, CDCI₃): δ = 1.98 (tt, 2H, J₃₋₄=6.5Hz, J₃₋₂=5Hz, H-3), 2.14 (s, 3H, CH₃), 2.27 (s, 3H, SCH₃), 2.73 (t, 2H, J₄₋₃=6.5Hz, H-4), 4.22 (t, 2H, J₂₋₃=5Hz, H-2), 6.75 (s, 1H, OH), 7.07 (bs. 1H, H-5). 13 C NMR (50 MHz, CDCI₃): δ = 8.78 (CH₃), 20.57 (S-CH₃), 22.32 (C-3), 24.46 (C-4), 66.57 (C-2), 110.93 (C-8), 111.27 (C-6), 114.37 (C-10), 132.73 (C-5), 153.46 (C-7), 154.78 (C-9).

Anal. Cald for C11H14O2S: C, 62.83; H, 6.71. Found: C, 62.74; H, 6.76.

General procedure for the preparation of annulation intermediates 3 and 9.

In a three-necked round bottomed flask equiped with a dropping funnel, a suspension of NaH (2 eq) in CH₂Cl₂ (1 mL/mmol NaH)was cooled at 0°C with an ice bath. Thioacetal 1a or 1b (1 eq) in CH₂Cl₂ (1 mL/mmol thioacetal) was added dropwise to the suspension. After stirring for 10 min at 0°C, α-methylenelactone 2 or 8 (1 eq) dissolved in CH₂Cl₂ (1 mL/mmol lactone) was added dropwise and the ice bath was replaced by a heating bath. The suspension was refluxed for 16h, allowed to stand and poured onto a 10% aqueous HCl solution (10 mL/mmol thioacetal). The organic layer was decanted, the aqueous layer extracted with CH₂Cl₂ and the joined organic layers washed with brine, dried over Na₂SO₄ and concentrated under vacuum. The residue was dissolved in acetic acid (5ml/mmol thioacetal) and the solution heated at 60°C during 1 hour. After cooling and distillation of acetic acid under vacuum, the product was submitted to flash-chromatography.

2,3,9,4-Tetrahydro-5,5-bis(methylthio)-6-one-benzofuran, 3a

Prepared from compounds 1a and 2. Chromatography: CH₂Cl₂/MeOH 99/1. The analytical sample was recristallized from CH₂Cl₂/hexane. Yield: 50%, white cristals, R_f : 0.27 (CH₂Cl₂/MeOH 99/1), mp: 111-111.5°C.

IR (CCl4): 2980-2820 (C-H), 1635 (C=O). 1 H NMR (200 MHz, CDCl3): δ = 1.83 (qd, 1H, J_{3ax-3eq}=12Hz, J_{3ax-2ax}=J_{3ax-9ax}=12Hz, J_{3ax-2eq}=8.5Hz, H-3ax), 2.07 (s, 3H, SCH3), 2.13 (s, 3H, SCH3), 2.28 (B part of ABX, 1H, J_{4ax-4eq}=13Hz, J_{4ax-9ax}=12Hz, H-4ax), 2.35 (ddd, 1H, J_{3eq-3ax}=12Hz, J_{3eq-2ax}=5Hz, J_{3eq-2eq}=0, J_{3eq-9ax}=8Hz, H-3eq), 2.60 (A part of ABX, 1H, J_{4eq-4ax}=13Hz, J_{4eq-9ax}=5Hz, H-4eq), 3.34 (dtd(d), 1H, J_{9ax-3ax}=12Hz, J_{9ax-3eq}=8Hz, J_{9ax-4ax}=12Hz, J_{9ax-4eq}=5Hz; J_{9ax-7}=2Hz, H-9), 4.30 (ddd, 1H; J_{2ax-2eq}=8.5Hz, J_{2ax-3ax}=12Hz, J_{2ax-3eq}=5Hz, H-2ax), 4.56 (t, 1H, J_{2eq-2ax}=J_{2eq-3ax}=8.5Hz, J_{2eq-3eq}=0, H-2eq), 5.43 (bs, 1H, H-7). 13 C NMR (50 MHz, CDCl₃): δ = 11.05 (SCH₃), 11.85 (SCH₃), 29.90 (C-3), 37.47 (C-9), 40.40 (C-4), 62.39 (C-5), 73.18 (C-2), 97.09 (C-7), 180.40 (C-8), 191.75 (C-6) Anal. Cald for C₁₀H₁₄O₂S₂: C, 52.14; H, 6.13. Found: C, 52.08; H, 5.91.

2,3,10,5-Tetrahydro-6,6-bis(methylthio)-7-one-benzopyran, 9a

Prepared from compounds **8** and **1a**. Chromatography: CH₂Cl₂/MeOH 99/1. The analytical sample was recristallized from CH₂Cl₂/hexane. Yield: 50%, white cristals, R_f: 0.29 (CH₂Cl₂/MeOH 99/1), mp: 102-104 °C.

IR (CCl₄): 2980-2840 (C-H), 1650 (C=O), 1615 (C=C). 1 H NMR (200 MHz, CDCl₃): δ = 1.41 (m, 1H, H-3 or H-4), 1.97 (m, 3H, H-3 and H-4), 2.05 (s, 3H, SCH₃), 2.07 (s, 3H, SCH₃), 2.32 (m, 2H, H-5), 2.85 (m, 1H, H-10), 3.97 (m, 1H, H-2), 4.29 (m, 1H, H-2), 5.47 (d, 1H, J₈₋₁₀=2Hz, H-8), 13 C NMR (50 MHz, CDCl₃): δ = 10.87 (SCH₃), 11.82 (SCH₃), 23.12 (C-4), 25.50 (C-3), 32.30 (C-10), 41.56 (C-5), 62.93 (C-6), 68.31 (C-2), 104.54 (C-8), 175.90 (C-9), 190.91 (C-7).

Anal. Cald for C11H16O2S2: C, 54.07; H, 6.60; Found: C, 53.85; H, 6.42.

2,3,9,4-Tetrahydro-5,5-bis(methylthio)-6-one-7-methyl-benzofuran. 3b

Prepared from compounds 1b and 2. Chromatography: CH₂Cl₂. Yield: 50%, white cristals. Rf: 0.50 (CH₂Cl₂/MeOH 99/1), mp: 98-99 °C

IR (CCl4) : 3000-2860 (C-H). 1675 (C=O). 1640 (C=C). 1 H NMR (200 MHz, CDCl3) : δ = 1.70 (bs, 3H, CH3). 1.78 (m, 1H, H-3ax), 2.04 (s, 3H, SCH3), 2.08 (s, 3H, SCH3), 2.27 (m, 2H, H-3eq + H-4ax), 2.55 (A part of ABX, 1H, J_{4eq}-4_{ax}=13Hz, J_{4eq}-9_{ax}=5Hz, H-4eq), 3.26 (m, 1H, H-9), 4.25 (ddd, 1H, J_{2ax-2eq}=8.5Hz, J_{2ax-3ax}=11.5Hz, J_{2ax-3eq}=5Hz, H-2ax), 4.53 (t, 1H, J_{2eq-2ax}=J_{2eq-3ax}=8.5Hz, H-2eq). 13 C NMR (50 MHz, CDCl3) : δ = 7.77 (CH3), 11.26 (SCH3), 11.99 (SCH3), 30.77 (C-3), 36.93 (C-9), 40.42 (C-4), 62.67 (C-5), 72.60 (C-2), 104.43 (C-7), 174.98 (C-8), 190.50 (C-6).

Anal. Cald for C11H16O2S2: C, 54.07; H, 6.60. Found: C, 53.88; H, 6.57.

2,3,10,5-Tetrahydro-6,6-bis(methylthio)-7-one-8-methyl-benzopyran, 9b

Prepared from compounds 8 and 1b. Chromatography: CH₂Cl₂ to CH₂Cl₂/MeOH 98/2 by 1%. Yield: 58%, white cristals. R_f: 0.43 (CH₂Cl₂/MeOH 99/1), mp: 106-107 °C.

IR (CCl4) : 3000-2860 (C-H). 1700 (C=O), 1650 (C=C). 1 H NMR (200 MHz, CDCl3) : δ = 1.39 (m, 1H, H-3 or H-4), 1.71 (bs, 3H, CH3), 1.93 (m, 3H, H-3 + H-4), 2.08 (bs, 6H, SCH3), 2.20 (m, 2H, H-5), 2.82 (m, 1H, H-10), 3.98 (m, 1H, H-2). 4.31 (m, 1H, H-2). 13 C NMR (50 MHz, CDCl3) : δ = 7.83 (CH3) , 10.96 (SCH3), 11.93 (SCH3), 23.28 (C-3), 26.48 (C-4), 32.03 (C-10), 41.18 (C-5), 63.05 (C-6), 68.03 (C-2), 111.38 (C-8), 169.83 (C-9), 190.91 (C-7).

Anal. Cald for C12H18O2S2: C, 55.78, H; 7.02. Found: C, 55.95; H, 6.99.

Transacetalization compounds

The cyclic dithioacetal 3a, 3b, 9a or 9b (1 eq) was dissolved in anhydrous CHCl₃ (5 mL/mmol thioacetal). Mercury acetate (2.5 eq) in MeOH (2 mL/mmol thioacetal) was added dropwise and the resulting solution stirred at room temperature for 16h. Unsoluble mercury salts were formed and the solution became pink. The solvents were removed under reduced pressure and the residue suspended in CH₂Cl₂ (5 mL/mmol thioacetal). The suspension was filtrated on celite to remove mercury salts. The clear filtrate was washed first with a 10% solution of sodium hydrogenosulfite, then with brine. The organic layer was decanted, dried over sodium sulfate and the solvent removed to give a yellow-brown wax. The crude product was purified by flash-chromatography on silica gel with the appropriate eluent (see products description).

2,3,9,4-Tetrahydro-5,5-bis(methoxy)-6-one-benzofuran, 5a

Prepared from compound 3a. Chromatography: $CH_2Cl_2/MeOH$ 98/2. Yield: 72%, a pale pink solid. $R_f: 0.25$ ($CH_2Cl_2/MeOH$ 98/2)

¹H NMR (200 MHz, CDCl₃): δ = 1.75 (m, 2H, H-3ax + H-4ax), 2.32 (m, 1H, H-3eq), 2.68 (A part of ABX, 1H, J_{4eq}-4_{ax}=12.5Hz, J_{4a}-9_{ax}=5Hz, H-4eq), 3.23 (s, 3H, OCH₃), 3.28 (m, 1H, H-9), 3.35 (s, 3H, OCH₃), 4.29 (ddd, 1H, J_{2ax}-2eq=8.5Hz, J_{2ax}-3_{ax}=12Hz, J_{2ax}-3eq=5Hz, H-2ax), 4.55 (t, 1H, J_{2eq}-2ax=J_{2eq}-3_{ax}=8.5Hz, H-2eq), 5.42 (bs, 1H, H-7). ¹³C NMR (50 MHz, CDCl₃): δ = 30.23 (C-3), 36.03 (C-4), 37.72 (C-9). 48.71 (OCH₃), 50.55 (OCH₃), 73.49 (C-2), 96.20 (C-5), 98.51 (C-7), 182.08 (C-8), 191.26 (C-6).

2,3,9,4-Tetrahydro-5,5-bis(methoxy)-6-one-7-methyl-benzofuran, 5b

Prepared from compound **3b**. Chromatography: ethyl acetate/hexane 1/1. Yield: 76%, pale pink cristals. The analytical sample was recritallized in ethyl acetate/hexane to yield white cristals. R_f : 0.31 (CH₂Cl₂/MeOH 95/5), mp: 97-98 °C.

IR (CCl4) : 3040-2860 (C-H), 1675 (C=O), 1650 (C=C). 1 H NMR (200 MHz, CDCl3) : δ = 1.64-1.88 (m, 2H, H-3ax + H-4ax), 1.72 (bs, 3H, CH3), 2.32 (m, 1H, H-3eq), 2.66 (A part of ABX, 1H, J4eq-4ax=12.5Hz, J4eq-9ax=5Hz, H-4eq), 3.16 (m, 1H, H-9), 3.21 (s, 3H, OCH3), 3.35 (s, 1H, OCH3), 4.29 (ddd, 1H, J2ax-2eq=8.5Hz, J2ax-3ax=12Hz, J2ax-3eq=5Hz; H-2ax), 4.55 (t, 1H, J2eq-2ax=J2eq-3ax=8.5Hz, H-2eq); 13 C NMR (50 MHz, CDCl3) : δ = 7.19 (CH3), 30.64 (C-3), 35.88 (C-4), 36.47 (C-9), 48.58 (OCH3), 50.24 (OCH3), 72.51 (C-2), 95.84 (C-5), 105.49 (C-7), 176.39 (C-8), 190.61 (C-6).

Anal. Cald for C11H16O4: C, 62.25; H, 7.60. Found: C, 62.40; H, 7.82.

2,3,10,5-Tetrahydro-6,6-bis(methoxy)-7-one-8-methyl-benzopyran, 11b

Prepared from compound **9b**. Chromatography: $CH_2Cl_2/MeOH$ 98/2. Yield: 83%, a pale pink wax. R_f : 0.25 ($CH_2Cl_2/MeOH$ 95/5).

IR (CCl4) : 3000-2840 (C-H), 1675 (C=O), 1630 (C=C). 1 H NMR (200 MHz, CDCl3) : δ = 1.33 (m, 1H, H-4ax), 1.56 (A part of ABX, 1H, J_{5ax-5eq}=13Hz, J_{5ax-10ax}=4.5Hz, H-5ax,), 1.65 (bs, 3H, CH3), 1.88 (m, 3H, H-3 + H-4eq), 2.36 (B part of ABX, 1H, J_{5eq-5ax}=13Hz; J_{5eq-10ax}=2.5Hz, H-5eq)), 2.74 (m, 1H, H-10), 3.10 (s, 3H, OCH3), 3.28 (s, 1H, OCH3), 3.96 (m, 1H, H-2ax), 4.26 (m, 1H, H-2eq). 13 C NMR (50 MHz, CDCl3) : δ = 7.43 (CH3), 23.21 (C-4), 26.56 (C-3), 31.72 (C-10), 36.78 (C-5), 48.58 (OCH3), 50.47 (OCH3), 67.87 (C-2), 96.14 (C-6), 112.34 (C-8), 171.80 (C-9), 191.50 (C-7).

Anal. Cald for C12H18O4: C, 63.70; H, 8.02. Found: C, 63.70; H, 7.96.

Preparation of catechol monoethers

Dimethoxyacetal **5a**, **5b** or **11b** was dissolved in acetic acid (10 mL/mmol acetal) and the solution was refluxed during 2h. After removal of the solvent under reduced pressure, the residue was purified by flash-chromatography with the appropriate eluent (see products description).

2,3-Dihydro-5-methoxy-6-hydroxy-benzofuran, 6a

Prepared from compound 5a. Chromatography : CH₂Cl₂. Yield : 66%, a colorless oil. R_f : 0.52 (CH₂Cl₂/MeOH 95/5)

IR (CCl4): 3560 (O-H), 3050-2840 (C-H). 1 H NMR (200 MHz, CDCl3): δ = 3.14 (t, 2H, J=8.5Hz, H-3), 3.84 (s, 3H, OCH3), 4.54 (t, 2H, J=8.5Hz, H-2), 5.62 (bs, 1H, OH), 6.46 (s, 1H, H-7), 6.75 (s, 1H, H-4). 13 C NMR (50 MHz, CDCl3): δ = 29.90 (C-3), 57.07 (OCH3), 71.55 (C-2), 97.21 (C-7), 108.36 (C-4), 116.54 (C-9), 140.76 (C-6), 145.65 (C-5), 154.41 (C-8).

Anal. Cald for C9H10O3: C, 65.05; H, 6.07. Found: C, 65.33; H, 6.24.

2,3-Dihydro-5-methoxy-6-hydroxy-7-methyl-benzofuran, 6b

Prepared from compound **5b**. Chromatography: $CH_2Cl_2/hexane 1/1$. Yield: 36%, an unstable oil which becomes red. R_f : 0.23 ($CH_2Cl_2/hexane 1/1$).

IR (CCl₄): 3550 (O-H), 3020-2840 (C-H). 1 H NMR (200 MHz, CDCl₃): δ = 2.15 (s, 3H, CH₃), 3.15 (t, 2H, J=8.5Hz, H-3), 3.83 (s, 3H, OCH₃), 4.53 (t, 2H, J=8.5Hz, H-2), 5.70 (bs, 1H, OH), 6.63 (s, 1H, H-4).

2,3-Dihydro-6-methoxy-7-hydroxy-8-methyl-benzopyran, 12b

Prepared from compound **11b**. Chromatography: CH₂Cl₂. Yield: 36%, a pale yellow oil. R_f: 0.25 (CH₂Cl₂ IR (CCl₄): 3560 (O-H), 3000-2840 (C-H); ¹H NMR (200 MHz, CDCl₃): δ = 1.99 (tt, 2H, J₃₋₂=5Hz, J₃₋₄=6.5Hz, H-3), 2.13 (s, 3H, CH₃), 2.73 (t, 2H, J₄₋₃=6.5Hz, H-4), 3.83 (s, 3H, OCH₃), 4.17 (t, 2H, J₂₋₃=5Hz, H-2), 5.65 (bs, 1H, OH), 6.41 (s, 1H, H-5). ¹³C NMR (50 MHz, CDCl₃): δ = 8.21 (CH₃), 22.71 (C-4), 24.80 (C-3), 56.33 (OCH₃), 66.27 (C-2), 108.52 (C-5). 111.59 (C-8), 140.23 (C-7), 142.47 (C-6) Anal. Cald for C₁1H₁4O₃: C, 68.02, H, 7.27. Found: C, 67.92; H, 7.36.

Addition of methyllithium to cyclic acetals

- 1) Cyclic acetal **5b** or **11b** (1 eq) dissolved in anhydrous THF (10 mL/mmol acetal) was cooled at -78°C. Methyl lithium [1,6M/ether (2 eq)] was added dropwise. Starting material disappearance was monitored by TLC. Workup was performed by addition of a saturated ammonium chloride solution (10 mL/mmol acetal) and extraction of the medium with CH₂Cl₂ (3 x 10 mL/mmol acetal). The combined organic layers were dried over sodium sulfate and concentrated under vacuum to yield an unstable colorless oil which was used without further purification in the next step.
- 2) The preceeding alcohols were aromatised in acetic acid (10 mL/mmol alcohol) by heating at 60°C (oil bath temperature) during 1h. The acid was removed under reduced pressure and the residue purified by flash-chromatography with the appropriate eluent.

2,3-Dihydro-5-methoxy-6,7-dimethyl-benzofuran, 7b.

Prepared from compound 5b. Chromatography : CH₂Cl₂/hexane 1/1. Yield : 69.5%, a pale yellow oil. Rf : 0.48 (CH₂Cl₂/hexane 1/1)

IR (CCl4) : 3000-2840 (C-H). 1 H NMR (200 MHz, CDCl3) : δ = 2.14 (s, 3H, CH3), 2.15 (s, 3H, CH3), 3.20 (t, 2H, J=8.5Hz, H-3), 3.78 (s, 3H, OCH3), 4.53 (t, 2H, J=8.5Hz, H-2), 6.66 (s, 1H, H-4). 13 C NMR (50 MHz, CDCl3) : δ = 11.76 (CH3), 12.27 (CH3), 30.77 (C-3), 56.65 (OCH3), 70.58 (C-2), 105.77 (C-4), 119.20 (C-9), 122.61 (C-6), 124.79 (C-7), 151.95 (C-5), 152.65 (C-8).

2,3-Dihydro-6-methoxy-7,8-dimethyl-benzopyran, 13b

Prepared from compound **11b**. Chromatography: CH₂Cl₂/hexane 1/1. Yield: 64% (80/20 mixture of ether **13b** and the corresponding phenol), a pale yellow oil. R_f: 0.26 (CH₂Cl₂/hexane 1/1) IR (CCl₄): 3010-2840 (C-H). ¹H NMR (200 MHz, CDCl₃): δ = 1.99 (tt, 2H, J₃₋₂=5Hz, J₃₋₄=6.5Hz, H-3), 2.14 (bs, 6H, CH₃), 2.79 (t, 2H, J₄₋₃=6.5Hz, H-4), 3.77 (s, 3H, OCH₃), 4.18 (t, 2H, J₂₋₃=5Hz, H-2), 6.43 (s, 1H, H-5).

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